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CHAPTER 2—SCIENCE, MATTER, ENERGY, AND SYSTEMS

MULTIPLE CHOICE

- 1. In 1963, Bormann and Likens compared the output of two river valleys, one forested and the other clear cut. What were their findings?
 - a. The deforested valley had higher water flow and a decrease in nutrient loss.
 - b. The forested valley had higher water flow and a decrease in nutrient loss.
 - c. The forested valley had lower water flow and increase in nutrient loss.
 - d. The deforested valley had lower water flow and increase in nutrient loss.
 - e. The deforested valley had higher water flow and increase in nutrient loss.

ANS: E PTS: 1

TOP: Core Case Study: How Do Scientists Learn About Nature? Experimenting with a Forest

KEY: Bloom's: Remember

- 2. What is the definition of a scientific hypothesis?
 - a. A simulation of a system being studied
 - b. A possible explanation for an observation or experimentation
 - c. The data needed to answer a question
 - d. Procedures carried out under controlled conditions to gather information
 - e. A widely accepted theory

ANS: B PTS: 1 TOP: 2-1 What Do Scientists Do?

KEY: Bloom's: Remember NOT: Modified

- 3. When an overwhelming body of observations and measurements supports a scientific hypothesis or group of related hypotheses, it becomes a(n) _____.
 - a. hypothesis
 - b. scientific law
 - c. scientific variable
 - d. scientific theory
 - e. conclusion

ANS: D PTS: 1 TOP: 2-1 What Do Scientists Do?

KEY: Bloom's: Remember

- 4. What is a well-tested and widely accepted description of what scientists find happening repeatedly in nature in the same way?
 - a. theory
 - b. scientific law
 - c. hypothesis
 - d. conclusion
 - e. model

ANS: B PTS: 1 TOP: 2-1 What Do Scientists Do?

KEY: Bloom's: Remember NOT: Modified

- 5. What is the correct order of applying the scientific process to a problem?
 - a. hypothesis-question-observation-experimentation-conclusion-analysis
 - b. hypothesis-conclusion-question-observation-experimentation-analysis
 - c. observation-hypothesis-conclusion-experimentation-analysis-question
 - d. observation-question-hypothesis-experimentation-analysis-conclusion

	e. nypothesis-experimentation-observation-analysis-question-conclusion
	ANS: D PTS: 1 TOP: 2-1 What Do Scientists Do? KEY: Bloom's: Remember
6.	Which of the following is an example of an organic compound? a. H_2O b. $NaCl$ c. H_2SO_4 d. N_2O e. CH_4
	ANS: E PTS: 1 TOP: 2-2 What Is Matter and What Happens When It Undergoes Change? KEY: Bloom's: Understand NOT: Modified
7.	Complex carbohydrates are a type of a. lipid b. chemical formula c. monomer d. protein e. organic polymer
	ANS: E PTS: 1 TOP: 2-2 What Is Matter and What Happens When It Undergoes Change? KEY: Bloom's: Remember NOT: New
8.	What is the distinct piece of DNA containing instructions for making proteins? a. chromosome b. nucleotide c. amino acid d. cell membrane e. hydrocarbon
	ANS: C PTS: 1 TOP: 2-2 What Is Matter and What Happens When It Undergoes Change? KEY: Bloom's: Remember
9.	What is the fundamental structural and functional unit of life? a. atom b. macromolecule c. DNA d. cell e. organism
	ANS: D PTS: 1 TOP: 2-2 What Is Matter and What Happens When It Undergoes Change? KEY: Bloom's: Remember
10.	Thousands of genes make up a single a. chromosome b. DNA c. cell nucleus d. trait e. organism

	ANS: A PTS: 1 TOP: 2-2 What Is Matter and What Happens When It Undergoes Change? KEY: Bloom's: Remember NOT: New
11.	 Which statement is an example of a chemical change? a. Confetti is cut from pieces of paper. b. Water evaporates from a lake. c. Ice cubes are formed in the freezer. d. A plant converts carbon dioxide into carbohydrates. e. A tree is cut down in the forest.
	ANS: D PTS: 1 TOP: 2-2 What Is Matter and What Happens When It Undergoes Change? KEY: Bloom's: Apply NOT: Modified
12.	Radioactive decay is best characterized as a type of a. physical change b. nuclear change c. chemical change d. chemical decay e. organic change
	ANS: B PTS: 1 TOP: 2-2 What Is Matter and What Happens When It Undergoes Change? KEY: Bloom's: Remember NOT: New
13.	What law states that when matter undergoes a physical or chemical change, no atoms are created or destroyed? a. The second law of thermodynamics b. The law of conservation of matter c. The first law of thermodynamics d. The atomic exchange law e. The law of conservation of energy
	ANS: B PTS: 1 TOP: 2-2 What Is Matter and What Happens When It Undergoes Change? KEY: Bloom's: Remember NOT: New
14.	Electromagnetic energy travels in a. waves b. packets c. nodes d. modules e. chunks
	ANS: A PTS: 1 TOP: 2-2 What Is Matter and What Happens When It Undergoes Change? KEY: Bloom's: Remember NOT: New
15.	Nuclear occurs when two nuclei are forced together. a. decay b. fission c. dissipation d. fusion e. equilibrium

	ANS: D PTS: 1 TOP: 2-2 What Is Matter and What Happens When It Undergoes Change? KEY: Bloom's: Remember NOT: New
16.	Matter is anything that a. has mass and takes up space b. has the capacity to do work c. can be changed in form d. can produce change e. moves mass
	ANS: A PTS: 1 TOP: 2-2 What Is Matter and What Happens When It Undergoes Change? KEY: Bloom's: Remember
17.	What is the most basic building block of matter? a. molecules b. compounds c. ions d. atoms e. minerals
	ANS: D PTS: 1 TOP: 2-2 What Is Matter and What Happens When It Undergoes Change? KEY: Bloom's: Remember NOT: Modified
18.	Fundamental types of matter that have unique sets of properties and cannot be broken down into simpler substances by chemical means are called a. mixtures b. compounds c. isotopes d. elements e. atoms
	ANS: D PTS: 1 TOP: 2-2 What Is Matter and What Happens When It Undergoes Change? KEY: Bloom's: Remember
19.	Which substance is a compound? a. water b. oxygen c. nitrogen d. hydrogen e. carbon
	ANS: A PTS: 1 TOP: 2-2 What Is Matter and What Happens When It Undergoes Change? KEY: Bloom's: Understand NOT: Modified
20.	Protons, neutrons, and electrons are all a. forms of energy b. equal in mass c. subatomic particles d. negative ions

	e. charged particles
	ANS: C PTS: 1 TOP: 2-2 What Is Matter and What Happens When It Undergoes Change? KEY: Bloom's: Remember
21.	The atomic number is the number of a. atoms in a molecule b. protons in an atom c. neutrons in a molecule d. electrons in an atom e. protons, electrons, and neutrons in an atom
	ANS: B PTS: 1 TOP: 2-2 What Is Matter and What Happens When It Undergoes Change? KEY: Bloom's: Remember
22.	The mass number of an atom is equal to the sum of the a. neutrons and isotopes b. neutrons and electrons c. neutrons and protons d. protons and electrons e. ions and isotopes
	ANS: C PTS: 1 TOP: 2-2 What Is Matter and What Happens When It Undergoes Change? KEY: Bloom's: Remember
23.	Isotopes are forms of an element that differ from one another by having different a. atomic numbers b. numbers of electrons c. numbers of protons d. mass numbers e. electrical charges
	ANS: D PTS: 1 TOP: 2-2 What Is Matter and What Happens When It Undergoes Change? KEY: Bloom's: Remember
24.	An ion has a net positive or negative a. proton b. isotope c. charge d. acid e. electron
	ANS: C PTS: 1 TOP: 2-2 What Is Matter and What Happens When It Undergoes Change? KEY: Bloom's: Remember NOT: New
25.	What describes the measurement of the concentration of hydrogen ions compared to the concentration of hydroxide ions in a solution? a. ionization b. pH c. alkalinity

	d. covalent bondinge. isotope
	ANS: B PTS: 1 TOP: 2-2 What Is Matter and What Happens When It Undergoes Change? KEY: Bloom's: Remember
26.	Fossil fuels are best characterized as a type of a. atomic particle b. renewable energy c. nonrenewable energy d. electromagnetic energy e. thermal energy
	ANS: C PTS: 1 TOP: 2-3 What is energy and what happens when it undergoes change? KEY: Bloom's: Remember NOT: New
27.	High-quality energy can best be characterized as a. fossilized b. pure c. electromagnetic d. kinetic e. concentrated
	ANS: E PTS: 1 TOP: 2-3 What Is Energy and What Happens When It Undergoes Change? KEY: Bloom's: Remember NOT: New
28.	Which law states that no energy can be created or destroyed? a. The first law of thermodynamics b. The second law of thermodynamics c. The law of conservation of matter d. The environmental exchange law e. The law of homeostasis
	ANS: A PTS: 1 TOP: 2-3 What Is Energy and What Happens When It Undergoes Change? KEY: Bloom's: Remember NOT: New
29.	Energy efficiency refers to a. how much energy we use b. how much energy is wasted c. how much heat is produced d. getting more work out of the energy we use e. getting more energy out of our work
	ANS: D PTS: 1 TOP: 2-3 What Is Energy and What Happens When It Undergoes Change? KEY: Bloom's: Remember NOT: New
30.	What is an example of low-quality energy? a. electricity b. heat in the ocean c. nuclear fission

	d. gasolinee. food
	ANS: B PTS: 1 TOP: 2-3 What Is Energy and What Happens When It Undergoes Change? KEY: Bloom's: Remember
31.	What percentage of the energy used to produce food for living organisms, and to heat the earth, comes from the sun? a. 10 b. 29 c. 49 d. 79 e. 99
	ANS: E PTS: 1 TOP: 2-3 What Is Energy and What Happens When It Undergoes Change? KEY: Bloom's: Remember
32.	What does the first law of thermodynamics tell us? a. Doing work always creates heat. b. Altering matter is the best source of energy. c. Energy cannot be recycled. d. Energy is neither created nor destroyed. e. Energy cannot be converted.
	ANS: D PTS: 1 TOP: 2-3 What Is Energy and What Happens When It Undergoes Change? KEY: Bloom's: Remember
33.	The matter and energy laws tell us that we can recycle a. both matter and energy b. neither matter nor energy c. matter but not energy d. energy but not matter e. nothing and everything
	ANS: C PTS: 1 TOP: 2-3 What Is Energy and What Happens When It Undergoes Change? KEY: Bloom's: Remember NOT: Modified
34.	The energy "lost" by a system is a. converted into an equal amount of matter b. equal to the energy the system creates c. converted to lower-quality energy d. returned to the system eventually e. converted to higher-quality energy
	ANS: C PTS: 1 TOP: 2-3 What Is Energy and What Happens When It Undergoes Change? KEY: Bloom's: Analyze NOT: Modified
35.	Scientists classify energy as either a. chemical or physical b. kinetic or mechanical

	c. potential or mechanicald. potential or kinetice. chemical or kinetic
	ANS: D PTS: 1 TOP: 2-3 What Is Energy and What Happens When It Undergoes Change? KEY: Bloom's: Understand
36.	Which of the following represents kinetic energy? a. water in a reservoir behind a dam b. a rock held in your hand c. chemical energy stored in food d. water in a stream e. light from the sun
	ANS: D PTS: 1 TOP: 2-3 What Is Energy and What Happens When It Undergoes Change? KEY: Bloom's: Remember NOT: New
37.	Heat is best characterized as a kind of energy. a. light b. potential c. kinetic d. nuclear e. low
	ANS: C PTS: 1 TOP: 2-3 What Is Energy and What Happens When It Undergoes Change? KEY: Bloom's: Remember NOT: New
38.	Scientists estimate that about of the energy used in the United States is unavoidably wasted. a. 5% b. 18% c. 55% d. 84% e. 96%
	ANS: D PTS: 1 TOP: 2-3 What Is Energy and What Happens When It Undergoes Change? KEY: Bloom's: Remember NOT: New
39.	Energy can be formally defined as a. the random motion of molecules b. the ability to do work and transfer heat c. a force that is exerted over some distance d. the movement of molecules e. the loss of matter
	ANS: B PTS: 1 TOP: 2-3 What Is Energy and What Happens When It Undergoes Change? KEY: Bloom's: Understand
40.	Which of the following best illustrates potential energy? a. the wind blowing b. water in a stream

	d. a car at the top of a hille. electricity
	ANS: D PTS: 1 TOP: 2-3 What Is Energy and What Happens When It Undergoes Change? KEY: Bloom's: Apply NOT: Modified
41.	Time delays in feedback systems allow changes in the environment to build slowly until the changes reach a(n) a. synergy point b. input stage c. throughput d. tipping point e. bioaccumulation point
	ANS: D PTS: 1 TOP: 2-4 What Are Systems and How Do They Respond to Change? KEY: Bloom's: Remember
42.	What are two or more processes interacting such that the combined effect is greater than the sum of the individual effects? a. homeostasis b. synergistic interaction c. negative feedback d. entropy e. time delay
	ANS: B PTS: 1 TOP: 2-4 What Are Systems and How Do They Respond to Change? KEY: Bloom's: Remember
43.	A community knows the effects of chemical X when it is used alone. They also know the same for chemical Z, so they set safe limits for use for both chemicals. When the chemicals are released at safe levels on the same day, there is a massive fish kill. What is the most likely explanation? a. homeostasis b. synergistic interaction c. negative feedback d. positive feedback e. entropy
	ANS: B PTS: 1 TOP: 2-4 What Are Systems and How Do They Respond to Change? KEY: Bloom's: Apply
14.	Human events that affect the environment are generally characterized by a. predictability as a result of population size b. many experiences leading to accurate generalizations c. long delays between events and responses d. obvious and immediate feedback e. negative feedback
	ANS: C PTS: 1 TOP: 2-4 What Are Systems and How Do They Respond to Change? KEY: Bloom's: Analyze NOT: Modified

c. steam

45.	What is considered to be the most essential component of a system? a. throughputs b. DNA c. energy d. matter e. light
	ANS: D PTS: 1 TOP: 2-4 What Are Systems and How Do They Respond to Change? KEY: Bloom's: Remember NOT: Modified
46.	One property of a system is that it a. functions in a regular and predictable manner b. is highly randomized in its processes c. cannot be accurately modeled d. consists solely of inputs and outputs e. exists only in models
	ANS: A PTS: 1 TOP: 2-4 What Are Systems and How Do They Respond to Change? KEY: Bloom's: Understand NOT: Modified
47.	Feedback causes in a system. a. change b. equilibrium c. chaos d. error e. noise
	ANS: A PTS: 1 TOP: 2-4 What Are Systems and How Do They Respond to Change? KEY: Bloom's: Remember NOT: New
48.	Which of the following illustrates a negative feedback loop? a. melting polar ice b. exponential population growth c. a thermostat maintaining a certain temperature in your house d. the greenhouse effect e. vegetation removed from a stream valley
	ANS: C PTS: 1 TOP: 2-4 What Are Systems and How Do They Respond to Change? KEY: Bloom's: Apply NOT: Modified
49.	What is the point of a fundamental shift in the behavior of a system? a. negative feedback b. positive feedback c. tipping point d. time delay e. synergistic point
	ANS: C PTS: 1 TOP: 2-4 What Are Systems and How Do They Respond to Change? KEY: Bloom's: Understand NOT: New

50.	What term describes the lack of system response system during a period of time? a. threshold level b. time delay c. tipping point d. negative feedback e. positive feedback
	ANS: B PTS: 1 TOP: 2-4 What Are Systems and How Do They Respond to Change? KEY: Bloom's: Remember NOT: New
TRUI	E/FALSE
51.	Scientists tend to be highly skeptical of new data, hypotheses, and models until they can be tested and verified.
	ANS: T PTS: 1 TOP: 2-1 What Do Scientists Do? KEY: Bloom's: Remember
52.	When someone says that evolution is not important, "after all, it's just a theory," it is probable that they do not understand how scientists use the term "theory."
	ANS: T PTS: 1 TOP: 2-1 What Do Scientists Do? KEY: Bloom's: Understand
53.	Tentative or frontier science is performed by amateur scientists whose work will never be accepted by their peers.
	ANS: F PTS: 1 TOP: 2-1 What Do Scientists Do? KEY: Bloom's: Remember
54.	Scientists can disprove things but they cannot prove anything absolutely, which means there is always some uncertainty in science.
	ANS: T PTS: 1 TOP: 2-1 What Do Scientists Do? KEY: Bloom's: Understand
55.	Atoms have a net positive electrical charge.
	ANS: F PTS: 1 TOP: 2-2 What Is Matter and What Happens When It Undergoes Change? KEY: Bloom's: Remember
56.	A chemical formula is a shorthand way of writing the symbols for atoms or ions in a compound.
	ANS: T PTS: 1 TOP: 2-2 What Is Matter and What Happens When It Undergoes Change? KEY: Bloom's: Remember
57.	Methane, a hydrocarbon, is considered an organic molecule even though it contains only one carbon atom.

	ANS: T PTS: 1 TOP: 2-2 What Is Matter and What Happens When It Undergoes Change? KEY: Bloom's: Understand
58.	How useful matter is to humans as a resource is determined by its concentration, availability for use, and its potential.
	ANS: T PTS: 1 TOP: 2-2 What Is Matter and What Happens When It Undergoes Change? KEY: Bloom's: Remember
59.	In a nuclear fission reaction, atoms are destroyed.
	ANS: F PTS: 1 TOP: 2-2 What Is Matter and What Happens When It Undergoes Change? KEY: Bloom's: Remember
60.	According to the law of conservation of matter, once trash decomposes in a landfill, we have completely gotten rid of the matter that made up the trash.
	ANS: F PTS: 1 TOP: 2-2 What Is Matter and What Happens When It Undergoes Change? KEY: Bloom's: Understand
61.	Energy cannot be recycled.
	ANS: T PTS: 1 TOP: 2-3 What Is Energy and What Happens When It Undergoes Change? KEY: Bloom's: Remember
62.	Burning coal demonstrates the conversion of energy from kinetic to potential.
	ANS: F PTS: 1 TOP: 2-3 What Is Energy and What Happens When It Undergoes Change? KEY: Bloom's: Understand
63.	Energy consumption does not mean the disappearance of energy; rather, it is the conversion of energy from one form to another with no net loss.
	ANS: T PTS: 1 TOP: 2-3 What Is Energy and What Happens When It Undergoes Change? KEY: Bloom's: Remember
64.	The scientific principles of sustainability show that everything we do affects someone or something in the environment in some way.
	ANS: T PTS: 1 TOP: 2-4 What Are Systems and How Do They Respond to Change? KEY: Bloom's: Understand
65.	A negative feedback loop causes a system to further change in the same direction.
	ANS: F PTS: 1 TOP: 2-4 What Are Systems and How Do They Respond to Change?

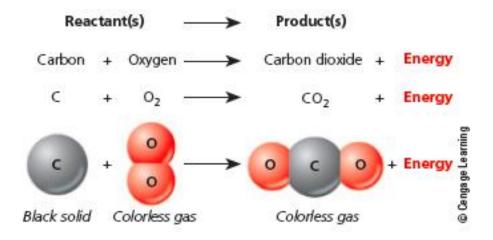
KEY: Bloom's: Understand

COMPLETION

66.	Science is based on the assumption that events in the natural world follow patterns that can be understood.
	ANS: cause-and-effect cause and effect
	PTS: 1 TOP: 2-1 What Do Scientists Do? KEY: Bloom's: Remember
67.	happens when scientists report details of their research and other scientists evaluate it.
	ANS: Peer review
	PTS: 1 TOP: 2-1 What Do Scientists Do? KEY: Bloom's: Remember
68.	A molecule is a combination of two or more atoms held together by forces called
	ANS: chemical bonds bonds
	PTS: 1 TOP: 2-2 What Is Matter and What Happens When It Undergoes Change? KEY: Bloom's: Remember
69.	Compounds are combinations of two or more different elements held together in proportions.
	ANS: fixed
	PTS: 1 TOP: 2-2 What Is Matter and What Happens When It Undergoes Change? KEY: Bloom's: Remember
70.	A(n) has more hydrogen ions than hydroxide ions and has a pH 7.
	ANS: acidic solution; less than
	PTS: 1 TOP: 2-2 What Is Matter and What Happens When It Undergoes Change? KEY: Bloom's: Remember
71.	An organic compound is one that contains one or more atoms combined with atoms of one or more other elements.
	ANS: carbon
	PTS: 1 TOP: 2-2 What Is Matter and What Happens When It Undergoes Change? KEY: Bloom's: Remember

72.	If a macromolecule was a brick wall it would be called a(n) ma repeating units called	made up of	
	ANS: polymer; monomers		
	PTS: 1 TOP: 2-2 What Is Matter and What Happens When It Undergoes KEY: Bloom's: Remember	Change?	
73.	Thousands of genes make up a single, a double helix DNA molecule w proteins.	rapped around	
	ANS: chromosome		
	PTS: 1 TOP: 2-2 What Is Matter and What Happens When It Undergoes KEY: Bloom's: Remember	Change?	
74.	According to the, when a physical or chemical change occurs, a created or destroyed.	no atoms are	
	ANS: law of conservation of matter		
	PTS: 1 TOP: 2-2 What Is Matter and What Happens When It Undergoes KEY: Bloom's: Remember	Change?	
75.	Body fat of a human or other animal is a type of energy.		
	ANS: potential		
	PTS: 1 TOP: 2-2 What Is Matter and What Happens When It Undergoes KEY: Bloom's: Remember	Change?	
76.	Most of the energy from burning a gallon of gasoline is lost asheat.	energy called	
	ANS: low-quality low quality		
	PTS: 1 TOP: 2-3 What Is Energy and What Happens When It Undergoes KEY: Bloom's: Remember	Change?	
77.	A(n) occurs when an output of matter, energy, or information is the system as an input and leads to changes in the system.	s fed back into	
	ANS: feedback loop		
	PTS: 1 TOP: 2-3 What Is Energy and What Happens When It Undergoes KEY: Bloom's: Remember	Change?	
78.	There are many types of electromagnetic radiation, each with a differentenergy content.	and	
	ANS: wavelength		

	PTS: 1 TOP: 2-3 What Is Energy and What Happens When It Undergoes Change? KEY: Remember	
79.	A(n) is a set of components that function and interact in some regular way.	
	ANS: system	
	PTS: 1 TOP: 2-4 What Are Systems and How Do They Respond to Change? KEY: Bloom's: Remember	
80.	Any process that increases or decreases a change to a system is called a(n)	
	ANS: feedback feedback loop	
	PTS: 1 TOP: 2-4 What Are Systems and How Do They Respond to Change? KEY: Bloom's: Remember	
SHO	RT ANSWER	
81.	Briefly describe the scientific process outlined in this chapter.	
	ANS: Identify a problem, find out what is known about the problem, ask a question to investigate, perform ar experiment to collect data, propose a hypothesis to explain the data, use the hypothesis to make projections that can be tested, test projections with further experiments, accept or revise hypothesis.	
	PTS: 1 TOP: 2-1 What Do Scientists Do? KEY: Bloom's: Remember NOT: New	
82.	Explain how a scientific law comes to be accepted.	
	ANS: A scientific law comes after a hypothesis is very well tested by several different scientists. The law is well-tested and widely accepted by the scientific community.	
	PTS: 1 TOP: 2-1 What Do Scientists Do? KEY: Bloom's: Understand NOT: New	
83.	Draw the basic chemical reaction of carbon and oxygen (C + O ₂)	
	ANS:	



PTS: 1 TOP: 2-2 What Is Matter and What Happens When It Undergoes Change?

KEY: Bloom's: Remember NOT: New

84. Explain how heat is a form of kinetic energy.

ANS:

Heat is a form of kinetic energy because it is the total kinetic energy of all moving atoms, ions, and molecules in an object. The atoms, ions, and molecules are all vibrating, and kinetic energy is the energy of movement.

PTS: 1 TOP: 2-3 What Is Energy and What Happens When It Undergoes Change?

KEY: Bloom's: Understand NOT: New

85. Give an example of a positive feedback loop and explain the process.

ANS:

Answers may vary.

Melting glaciers are positive feedback loops. As ice melts, there is less light-colored ice to reflect sunlight, so more is absorbed, which warms the air and causes more melt.

PTS: 1 TOP: 2-4 What Are Systems and How Do They Respond to Change?

KEY: Bloom's: Apply NOT: New

ESSAY

86. Explain how the human body is intimately connected to the two laws of thermodynamics.

ANS:

(pages 46-47) The first law of thermodynamics says that energy can neither be created nor destroyed, only transformed. This is the basis of the flow of energy from the sun through living systems on earth, including humans. We take in energy in the form of chemicals assembled by other living organisms and transform it, using the energy to do the many things required to live. If any living organism fails to take in and transform energy for their purposes, the second law of thermodynamics takes over. The second law says entropy (randomness or disorder) tends to increase in energy systems. In shortened form, the second law means we will die and decompose. Energy is required to keep a system functioning. When the system is no longer taking in energy, randomness or disorder will increase.

PTS: 1 OBJ: Critical Thinking

TOP: 2-3 What Is Energy and What Happens When It Undergoes Change?

KEY: Bloom's: Understand NOT: Modified

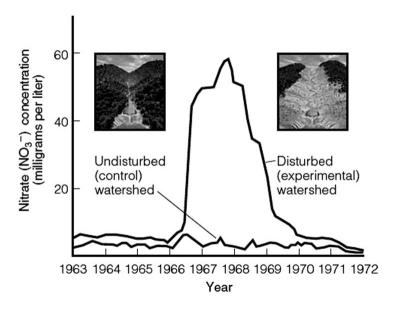
87. Differentiate between a hypothesis, a guess, and a theory. Explain why it is important for non-scientists to understand how scientists use these terms when discussing something like global warming or evolution. Why might it be incorrect when a non-scientist dismisses a topic like these as being "just a theory"?

ANS:

A hypothesis is an effort to explain phenomenon based on prior experience with the same or similar phenomena. It is often defined as an educated guess. The usual way to define a "guess" is the suggestion of an answer without prior experience. A theory is a structure intended to explain a series of phenomena, and is constructed from hypotheses that have been tested and not proven wrong. As such, a theory is based on substantial amounts of data.

PTS: 1 OBJ: Critical Thinking TOP: 2-1 What Do Scientists Do?

KEY: Bloom's: Understand



88. With reference to the accompanying figure, explain the significant difference that occurred between 1966 and 1970 in terms of the two lines representing the control watershed and the experimental watershed.

ANS:

The line labeled "Disturbed (experimental) watershed" represents the level of loss of nitrates following the removal of vegetation in the experimental watershed. Without vegetation the soil rapidly lost the nitrates to rainfall until they were substantially gone.

PTS: 1 OBJ: Critical Thinking

TOP: Figure 2-5 | 2-2 What Is Matter and What Happens When It Undergoes Change?

KEY: Bloom's: Analyze

89. Much of the energy produced is lost before it can become useful. Explain how energy efficiency, or energy productivity, and the second law of thermodynamics may be useful in a discussion with another person on how to reduce CO₂ and other greenhouse gas emissions.

ANS:

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Machines that use fossil fuels are very energy-inefficient, converting a small percentage of the energy in the fuel source to useful activities. An effort to increase the level of efficiency would substantially reduce the amount of fossil fuel that needed to be converted and would reduce the amount of emissions of CO₂ and other greenhouse gases.

PTS: 1 OBJ: Critical Thinking

TOP: 2-3 What Is Energy and What Happens When It Undergoes Change?

KEY: Bloom's: Understand

90. How is the concept of an environmental threshold or tipping point important in regards to global warming?

ANS:

A tipping point is a level at which a critical mass has been reached that causes an event to occur, an event that may be irreversible. If humans cause the climate of the earth to warm beyond a certain level, it may be impossible to correct the situation, and the climate may be irreversibly altered for the worse.

PTS: 1 OBJ: Critical Thinking

TOP: 2-4 What Are Systems and How Do They Respond to Change?

KEY: Bloom's: Understand